

Transmitted by the Informal group

Informal document No. GRPE-64-06

on Gaseous Fuelled Vehicles (GFV)

(64th GRPE, 5-8 June 2012, agenda items 3c and 8b)

Proposal for amendments to Documents

ECE-TRANS-WP29-GRPE-2012-07

and

ECE-TRANS-WP29-GRPE-2012-06

The text reproduced below was prepared by the Chair of the informal group on Gaseous Fuelled Vehicles (GFV) to correct a measurement error contained in docs. ECE-TRANS-WP29-GRPE-2012-06 (R83) and ECE-TRANS-WP29-GRPE-2012-07 (R.115) when weighing method and G25 reference fuel are used. The modifications to the original English text are marked in bold for new or strikethrough for deleted characters.

I. Proposal

Annex 6B of doc. ECE-TRANS-WP29-GRPE-2012-07 Paragraph 2., amend to read:

2. Calculation of the CNG energy ratio

The fuel consumption value shall be calculated from the emissions of hydrocarbons, carbon monoxide, and carbon dioxide determined from the measurement results assuming that only CNG is burned during the test.

The CNG ratio of the energy consumed in the cycle is then determined as follows:

$$G_{\text{CNG}} = M_{\text{CNG}} * \underline{\text{cf}} * 100 / (\text{FC}_{\text{mean}} * \text{dist} * d) * 100\%$$

Where:

G_{CNG} : the CNG energy ratio;

M_{CNG} : the CNG mass consumed during the cycle (kg);

FC_{mean} : the mean fuel consumption calculated in accordance with paragraph. 6.2.2.4.3.2.;

dist: distance travelled during the cycle (km);

d: density $d=0.654\text{kg/m}^3$

cf: correction factor, assuming the following values:

cf = 1 **in case of G₂₀ reference fuel**

cf = 0.78 **in case of G₂₅ reference fuel**

Annex 12 Appendix 2 of doc. ECE-TRANS-WP29-GRPE-2012-06 Paragraph 2., amend to read:

2. Calculation of the CNG energy ratio

The fuel consumption value shall be calculated from the emissions of hydrocarbons, carbon monoxide, and carbon dioxide determined from the measurement results assuming that only CNG is burned during the test.

The CNG ratio of the energy consumed in the cycle is then determined as follows:

$$G_{\text{CNG}} = M_{\text{CNG}} * \underline{\text{cf}} * 100 / (\text{FC}_{\text{norm}} * \text{dist} * d)$$

Where:

G_{CNG} : the CNG energy ratio;

M_{CNG} : the CNG mass consumed during the cycle (kg);

FC_{norm} : the fuel consumption calculated in accordance with Paragraph 1.4.3., letter (c), of Annex 6 to Regulation No. 101;

dist: distance travelled during the cycle (km);

d: density $d = 0.654\text{kg/m}^3$

cf: correction factor, assuming the following values:

cf = 1 **in case of G₂₀ reference fuel**

cf = 0.78 **in case of G₂₅ reference fuel**

II. Justification

Documents ECE-TRANS-WP29-GRPE-2012-07 (referring to R115) and ECE-TRANS-WP29-GRPE-2012-06 (referring to R83) introduce the calculation of NG/biomethane energy ratio during a type I cycle by weighing the NG/biomethane mass consumed. The formulas introduced respectively in Annex 6B (ECE-TRANS-WP29-GRPE-2012-07) and Annex 12 - Appendix 2 (ECE-TRANS-WP29-GRPE-2012-06) are affected by an error when G₂₅ reference fuel is used. The error is due to the presence of inert gas (N₂) in the G₂₅ reference fuel, varying from 12 to 16% in molar fraction units.

This proposal describes a modification to previous formulas in order to take into account the presence of inert gas and providing the correct energy ratio. When G₂₅ reference fuel is used, its composition is known and the mass of inert gas in it can be calculated.

R83 fixes the allowed ranges for both CH₄ molar fraction and N₂ molar fraction in G₂₅ reference fuel. If we assume x_{CH_4} the molar fraction of CH₄ and x_{N_2} the molar fraction of N₂, we have:

$$x_{CH_4} = n_{CH_4} / n_{TOT} \quad (\text{for } G_{25}: 0.84 \leq x_{CH_4} \leq 0.88)$$

$$x_{N_2} = n_{N_2} / n_{TOT} \quad (\text{for } G_{25}: 0.12 \leq x_{N_2} \leq 0.16)$$

where:

n_{CH_4} is the number of moles of CH₄

n_{N_2} is the number of moles of N₂

n_{TOT} is the total number of moles

Since the mass of CH₄ and N₂ expressed in grams can be calculated as follows:

$$m_{CH_4} = n_{CH_4} * PM_{CH_4}$$

$$m_{N_2} = n_{N_2} * PM_{N_2}$$

where:

PM_{CH_4} is the molar mass of CH₄ = 16.042 g/mol

PM_{N_2} is the molar mass of N₂ = 28.02 g/mol

Combining previous equations:

$$m_{CH_4} = x_{CH_4} * n_{TOT} * PM_{CH_4}$$

$$m_{N_2} = x_{N_2} * n_{TOT} * PM_{N_2}$$

The weighing process will measure the total mass of G₂₅ m_{TOT} :

$$m_{TOT} = m_{CH_4} + m_{N_2}$$

$$m_{TOT} = n_{TOT} * (x_{CH_4} * PM_{CH_4} + x_{N_2} * PM_{N_2})$$

but for the calculation of the energy ratio only the mass of CH₄ should be used. Thus the weighing process will overestimate the mass by a factor:

$$m_{TOT} / m_{CH_4} = (x_{CH_4} * PM_{CH_4} + x_{N_2} * PM_{N_2}) / (x_{CH_4} * PM_{CH_4})$$

or vice versa the measured mass must be reduced by a factor:

$$cf = m_{CH_4} / m_{TOT} = (x_{CH_4} * PM_{CH_4}) / (x_{CH_4} * PM_{CH_4} + x_{N_2} * PM_{N_2})$$

According to the allowed range for N₂ content, this correction factor will also vary:

- from 0.75, in case of minimum quantity of CH₄ (0.84) and maximum quantity of N₂ (0.16),

- to 0.8, in case of maximum quantity of CH₄ (0.88) and minimum quantity of N₂ (0.12)

Thus a mean value of 0.78 can be adopted for G₂₅ reference fuel.