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Study Report

WD-SI VA 1

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Thermal Properties of Azodicarbonamide Formulations

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Safety Assessment

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1. Assessment Task

A wide range of azodicarbonamide (AC) formulations were to be tested with respect to their thermal properties. The influence of their ingredients on the decomposition behavior was to be studied, and a correlation between the azodicarbonamide content and the decomposition energy was to be found.

2. Assessment

This is a short version (summary) of the study report for public purposes (presentation to the UN). All confidential information is disclosed.

The 22 formulations investigated cover a wide range of AC concentrations, activators and inert material (inorganic, organic, polymer). The overall decomposition energy was obtained from DSC measurements using the same machine (see below) and plotted against the known AC content. Activated and non-activated AC formulations were first treated separately.

A linear regression was performed on each group with the y intercept set to zero. The upper and lower confidence limits (95 % probability) were calculated and added to the plot.

The data obtained are as follows:

Correlation: (Decomposition energy) [J/g] = slope * (AC content) [%]

	Pure AC formulations	Activated AC formulations	All AC formulations
Observations	10	12	22
Multiple correlation coeff (R)	0,982	0,987	0,985
R²	0,965	0,974	0,971
Intercept (set)	0	0	0
Slope	11,8	12,0	11,9
Standard error (regression)	52,8	58,4	54,9
Analysis of variation (ANOVA)			
Sums of squares (regression)	684012	1413343	2099992
Sum of squares (residue)	25072	37507	63315
Aver. sum of quares (residue)	2786	3410	3015
Standard error (slope)	0,38	0,35	0,25
Lower 95 % (slope)	10,9	11,3	11,4
Upper 95 % (slope)	12,6	12,8	12,5

Since no significant differences between the 2 groups could be detected, they were united and a third regression calculation was performed. The results are given in the table. The corresponding plot is shown in diagram 3.

The regression line may be calculated from the formula

$$\text{(Decomp. energy) [J/g]} = 11,9 * \text{AC content [\%]}.$$

Upper (UCL) and lower (LCL) confidence limits are obtained from the following equation

$$\text{UCL} = \text{(Decomp. energy)} + 2,08 * (\text{cl})^{0,5}$$

$$\text{LCL} = \text{(Decomp. energy)} - 2,08 * (\text{cl})^{0,5} \quad , \text{ with}$$

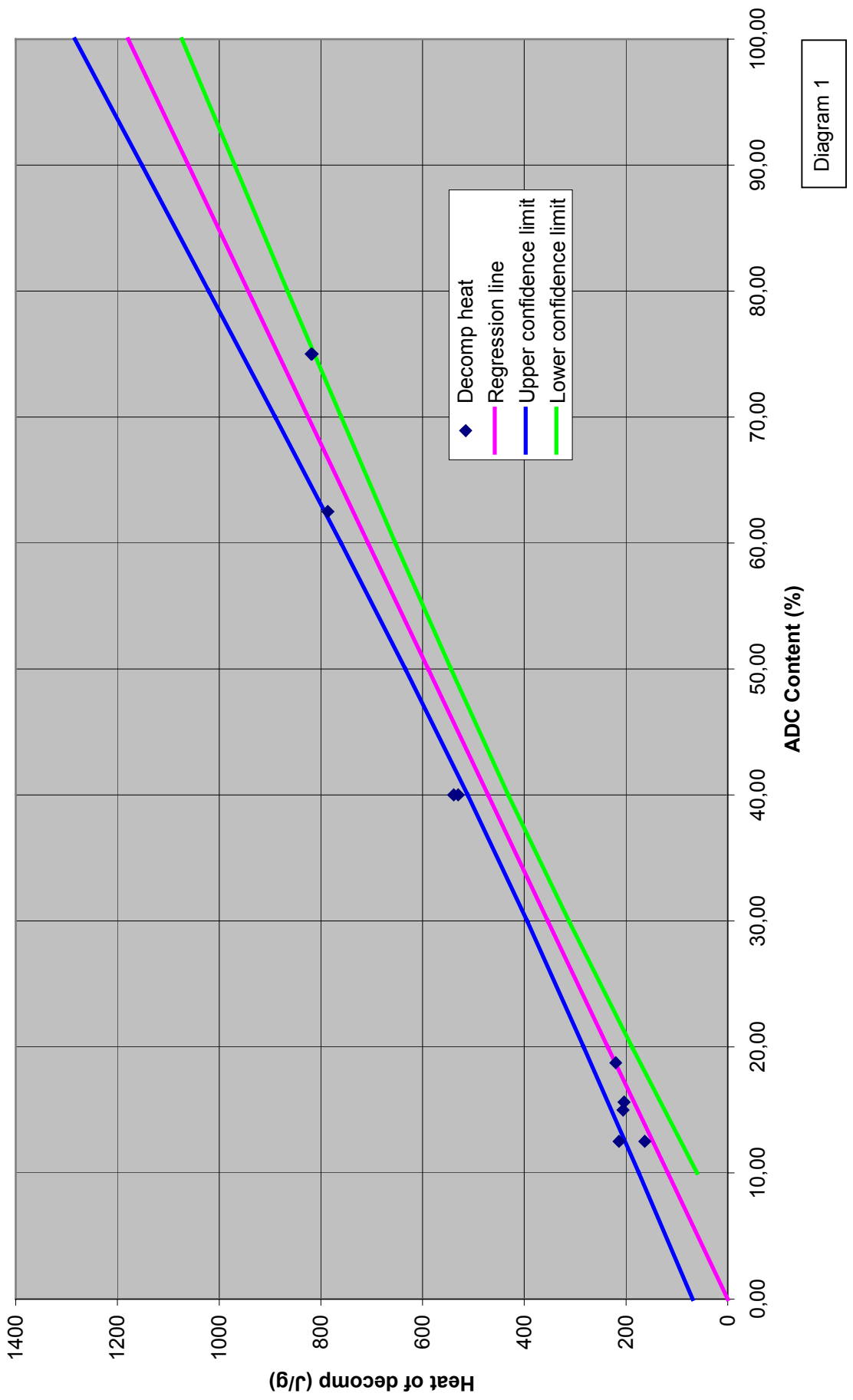
$$(\text{cl}) = 3015 * (1/21 + ((\text{AC content}) - 37,773)^2 / 16508,2)$$

From these data one may calculate that for AC formulations containing less than 40 % AC, the decomposition energy will not exceed 500 J/g. According to the UN Manual of Tests and Criteria, Appendix 6, paragraph 3.3, for such formulations the acceptance procedure for Class 1 explosives need not be applied.

3. Notes

All measurements were performed in Bayer's Process and Plant Safety Lab using a DSC instrument of Bayer's own design that had previously been carefully calibrated.

Pure ADC types



Activated types

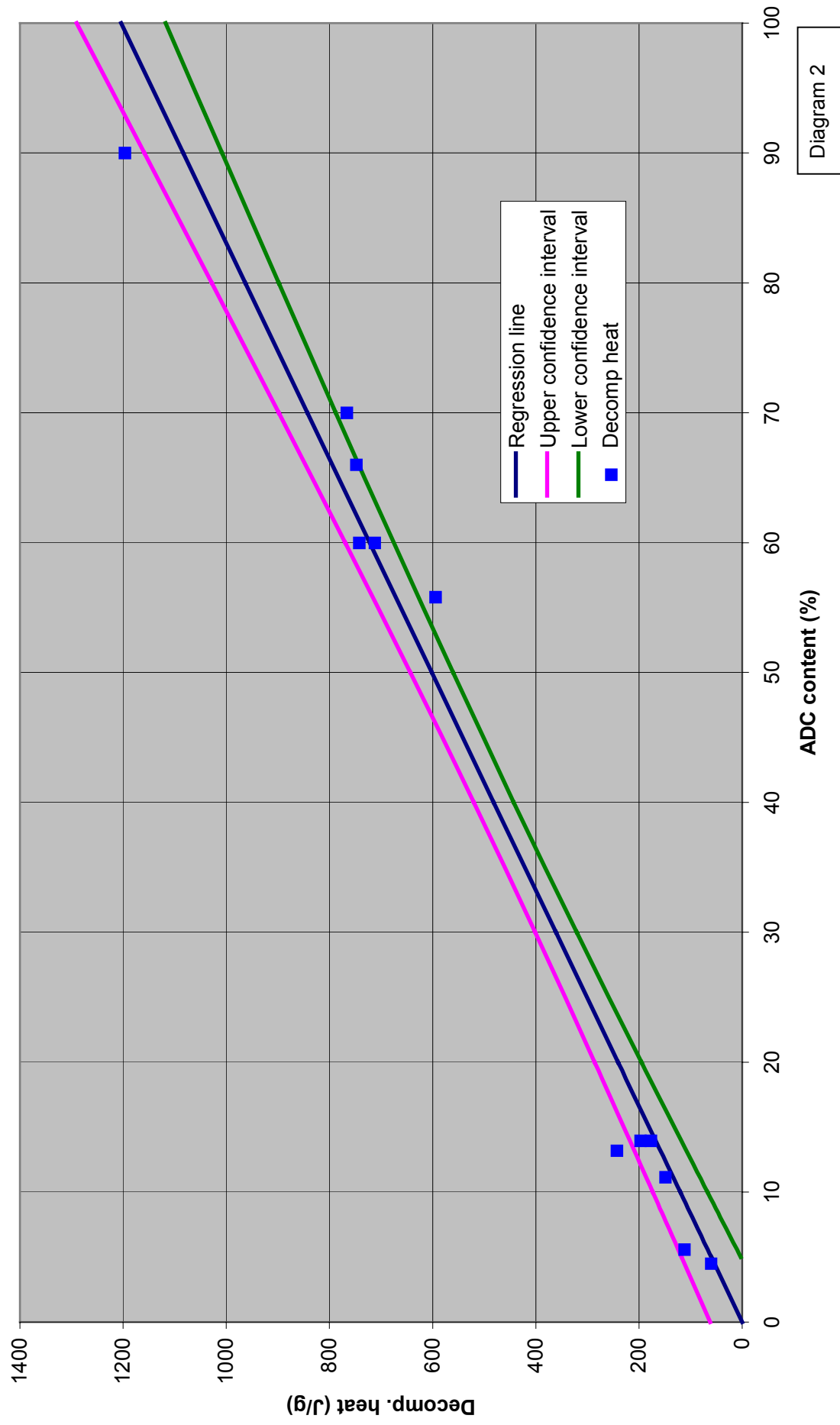


Diagram 2

Pure ADC + activated Types

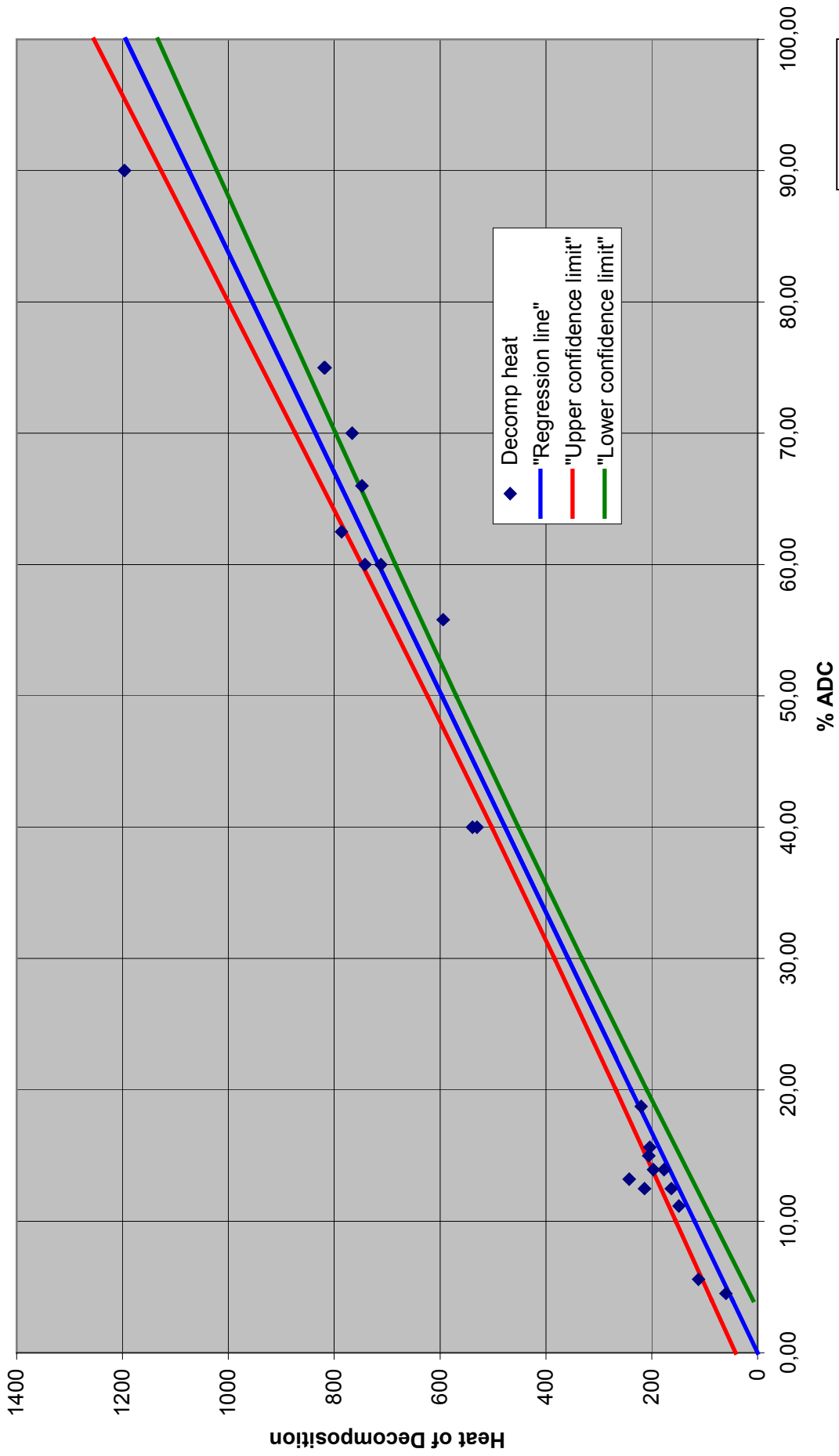


Diagram 3